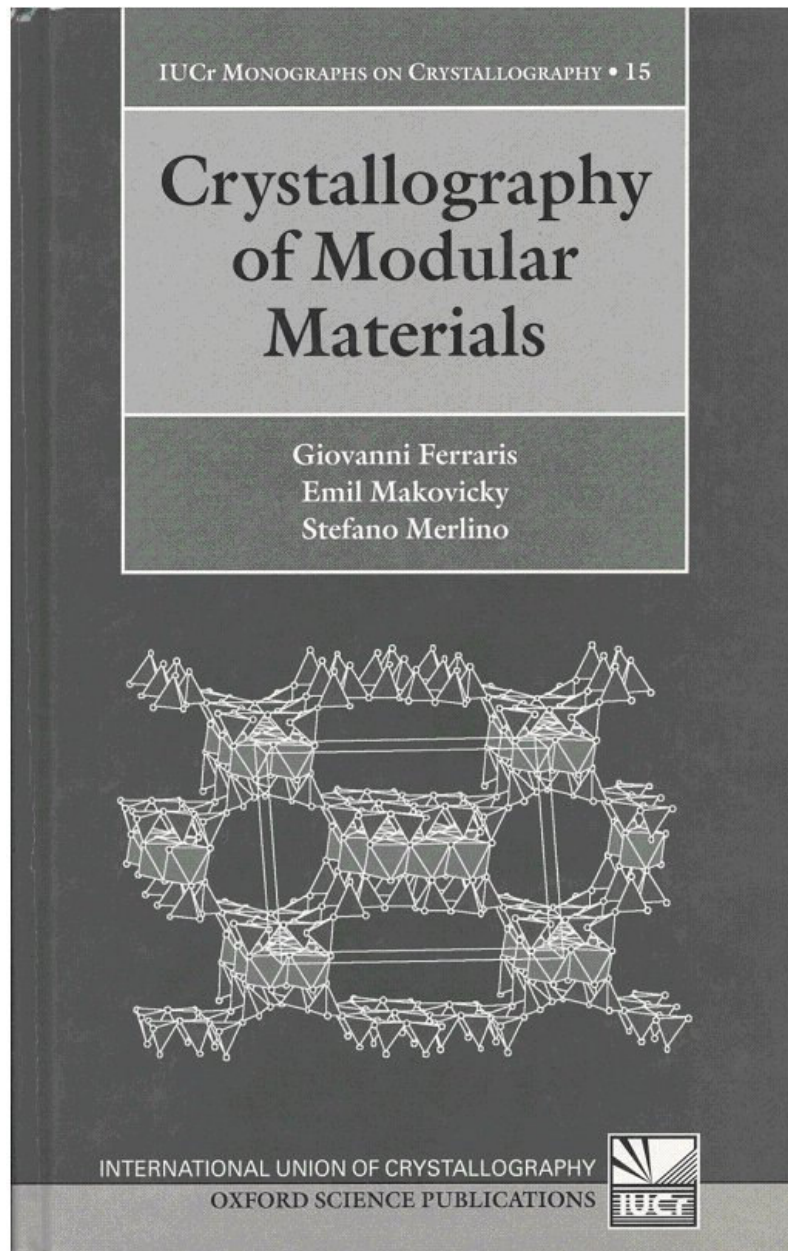


Modelling Modular Inorganic Structures

EXERCISES

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Exercise 1

CaF₂ (fluorite) cubic (Fig. 1s see solution)

- *Chemical composition:* CaF₂ - *Cell parameter:* $a = 5.462 \text{ \AA}$
- *Density:* $\delta = 3.18 \text{ gr/cm}^3$ - *Atomic weights:* Ca 40.08, F 18.998
- *Symmetry:* hkl diffractions occur only if h, k, l are either all odd or all even; from morphology the point group of fluorite is $m\bar{3}m$; then, the space group is $Fm\bar{3}m$. Assign atoms to the Wyckoff positions.

Number of positions, Wyckoff notation, and point symmetry	Co-ordinates of equivalent positions	Conditions limiting possible reflections		
	$(0,0,0; 0, \frac{1}{2}, \frac{1}{2}; \frac{1}{2}, 0, \frac{1}{2}; \frac{1}{2}, \frac{1}{2}, 0)+$			
		General:		
192	l	1	$x, y, z; z, x, y; y, z, x; x, z, y; y, x, z; z, y, x;$ $x, \bar{y}, \bar{z}; z, \bar{x}, \bar{y}; y, \bar{z}, \bar{x}; x, \bar{z}, \bar{y}; y, \bar{x}, \bar{z}; z, \bar{y}, \bar{x};$ $\bar{x}, y, \bar{z}; \bar{z}, x, \bar{y}; \bar{y}, z, \bar{x}; \bar{x}, z, \bar{y}; \bar{y}, x, \bar{z}; \bar{z}, y, \bar{x};$ $\bar{x}, \bar{y}, z; \bar{z}, \bar{x}, y; \bar{y}, \bar{z}, x; \bar{x}, \bar{z}, y; \bar{y}, \bar{x}, z; \bar{z}, \bar{y}, x;$ $\bar{x}, \bar{y}, \bar{z}; \bar{z}, \bar{x}, \bar{y}; \bar{y}, \bar{z}, \bar{x}; \bar{x}, \bar{z}, \bar{y}; \bar{y}, \bar{x}, \bar{z}; \bar{z}, \bar{y}, \bar{x};$ $\bar{x}, y, z; \bar{z}, x, y; \bar{y}, z, x; \bar{x}, z, y; \bar{y}, x, z; \bar{z}, y, x;$ $x, \bar{y}, z; z, \bar{x}, y; y, \bar{z}, x; x, \bar{z}, y; y, \bar{x}, z; z, \bar{y}, x;$ $x, y, \bar{z}; z, x, \bar{y}; y, z, \bar{x}; x, z, \bar{y}; y, x, \bar{z}; z, y, \bar{x}.$	$hkl: h+k, k+l, (l+h)=2n$ $hhl: (l+h=2n); \text{C}$ $okl: (k, l=2n); \text{C}$
96	k	m	$x, x, z; z, x, x; x, z, x; \bar{x}, \bar{x}, \bar{z}; \bar{z}, \bar{x}, \bar{x}; \bar{x}, \bar{z}, \bar{x};$ $x, \bar{x}, \bar{z}; z, \bar{x}, \bar{x}; x, \bar{z}, \bar{x}; \bar{x}, x, z; \bar{z}, x, x; \bar{x}, z, x;$ $\bar{x}, x, \bar{z}; \bar{z}, x, \bar{x}; \bar{x}, z, \bar{x}; x, \bar{x}, z; z, \bar{x}, x; x, \bar{z}, x;$ $\bar{x}, \bar{x}, z; \bar{z}, \bar{x}, x; \bar{x}, \bar{z}, x; x, x, \bar{z}; z, x, \bar{x}; x, z, \bar{x}.$	} no extra conditions
96	j	m	$0, y, z; z, 0, y; y, z, 0; 0, z, y; y, 0, z; z, y, 0;$ $0, \bar{y}, \bar{z}; \bar{z}, 0, \bar{y}; \bar{y}, \bar{z}, 0; 0, \bar{z}, \bar{y}; \bar{y}, 0, \bar{z}; \bar{z}, \bar{y}, 0;$ $0, y, \bar{z}; \bar{z}, 0, y; y, \bar{z}, 0; 0, \bar{z}, y; y, 0, \bar{z}; \bar{z}, y, 0;$ $0, \bar{y}, z; z, 0, \bar{y}; \bar{y}, z, 0; 0, z, \bar{y}; \bar{y}, 0, z; z, \bar{y}, 0.$	
48	i	mm	$\frac{1}{2}, x, x; x, \frac{1}{2}, x; x, x, \frac{1}{2}; \frac{1}{2}, x, \bar{x}; \bar{x}, \frac{1}{2}, x; x, \bar{x}, \frac{1}{2};$ $\frac{1}{2}, \bar{x}, \bar{x}; \bar{x}, \frac{1}{2}, \bar{x}; \bar{x}, \bar{x}, \frac{1}{2}; \frac{1}{2}, \bar{x}, x; x, \frac{1}{2}, \bar{x}; \bar{x}, x, \frac{1}{2}.$	
48	h	mm	$0, x, x; x, 0, x; x, x, 0; 0, x, \bar{x}; \bar{x}, 0, x; x, \bar{x}, 0;$ $0, \bar{x}, \bar{x}; \bar{x}, 0, \bar{x}; \bar{x}, \bar{x}, 0; 0, \bar{x}, x; x, 0, \bar{x}; \bar{x}, x, 0.$	
48	g	mm	$x, \frac{1}{4}, \frac{1}{4}; \frac{1}{4}, x, \frac{1}{4}; \frac{1}{4}, \frac{1}{4}, x; x, \frac{1}{4}, \frac{3}{4}; \frac{3}{4}, x, \frac{1}{4}; \frac{1}{4}, \frac{3}{4}, x;$ $\bar{x}, \frac{1}{4}, \frac{1}{4}; \frac{1}{4}, \bar{x}, \frac{1}{4}; \frac{1}{4}, \frac{1}{4}, \bar{x}; \bar{x}, \frac{1}{4}, \frac{3}{4}; \frac{3}{4}, \bar{x}, \frac{1}{4}; \frac{1}{4}, \frac{3}{4}, \bar{x}.$	$hkl: h, (k, l)=2n$
32	f	$3m$	$x, x, x; x, \bar{x}, \bar{x}; \bar{x}, x, \bar{x}; \bar{x}, \bar{x}, x;$ $\bar{x}, \bar{x}, \bar{x}; \bar{x}, x, x; x, \bar{x}, x; x, x, \bar{x}.$	} no extra conditions
24	e	$4mm$	$x, 0, 0; 0, x, 0; 0, 0, x; \bar{x}, 0, 0; 0, \bar{x}, 0; 0, 0, \bar{x}.$	} $hkl: h, (k, l)=2n$
24	d	mmm	$0, \frac{1}{4}, \frac{1}{4}; \frac{1}{4}, 0, \frac{1}{4}; \frac{1}{4}, \frac{1}{4}, 0; 0, \frac{1}{4}, \frac{3}{4}; \frac{3}{4}, 0, \frac{1}{4}; \frac{1}{4}, \frac{3}{4}, 0.$	
8	c	$\bar{4}3m$	$\frac{1}{2}, \frac{1}{2}, \frac{1}{2}; \frac{3}{4}, \frac{3}{4}, \frac{3}{4}.$	} no extra conditions
4	b	$m3m$	$\frac{1}{2}, \frac{1}{2}, \frac{1}{2}.$	
4	a	$m3m$	$0, 0, 0.$	

Exercise 2

MgAl₂O₄ (spinel) cubic (Fig. 2s, see solution)

- *Chemical composition:* MgAl₂O₄ - *Cell parameter:* $a = 8.09 \text{ \AA}$ - *Density:* $\delta = 3.58 \text{ gr/cm}^3$
- *Atomic weights:* Mg 24.312, Al 26.982, O 15.999 - *Space group:* hkl diffractions occur only if h, k, l are either all odd or all even: thus, lattice is cF ; besides, $0kl$ occur only for $k+l=4n$; from morphology the point group is $m\bar{3}m$, thus the space group is $Fd\bar{3}m$. Assign atoms to the Wyckoff positions.

	Number of positions, Wyckoff notation, and point symmetry	Co-ordinates of equivalent positions	Conditions limiting possible reflections
		$(0,0,0; 0, \frac{1}{2}, \frac{1}{2}; \frac{1}{2}, 0, \frac{1}{2}; \frac{1}{2}, \frac{1}{2}, 0) +$	
192	<i>i</i> 1	$x, y, z; z, x, y; y, z, x; x, z, y; y, x, z; z, y, x;$ $x, \bar{y}, \bar{z}; z, \bar{x}, \bar{y}; y, \bar{z}, \bar{x}; x, \bar{z}, \bar{y}; y, \bar{x}, \bar{z}; z, \bar{y}, \bar{x};$ $\bar{x}, y, \bar{z}; \bar{z}, x, \bar{y}; \bar{y}, z, \bar{x}; \bar{x}, z, \bar{y}; \bar{y}, x, \bar{z}; \bar{z}, y, \bar{x};$ $\bar{x}, \bar{y}, z; \bar{z}, \bar{x}, y; \bar{y}, \bar{z}, x; \bar{x}, \bar{z}, y; \bar{y}, \bar{x}, z; \bar{z}, \bar{y}, x;$ $\frac{1}{4}-x, \frac{1}{4}-y, \frac{1}{4}-z; \frac{1}{4}-z, \frac{1}{4}-x, \frac{1}{4}-y; \frac{1}{4}-y, \frac{1}{4}-z, \frac{1}{4}-x;$ $\frac{1}{4}-x, \frac{1}{4}+y, \frac{1}{4}+z; \frac{1}{4}-z, \frac{1}{4}+x, \frac{1}{4}+y; \frac{1}{4}-y, \frac{1}{4}+z, \frac{1}{4}+x;$ $\frac{1}{4}+x, \frac{1}{4}-y, \frac{1}{4}+z; \frac{1}{4}+z, \frac{1}{4}-x, \frac{1}{4}+y; \frac{1}{4}+y, \frac{1}{4}-z, \frac{1}{4}+x;$ $\frac{1}{4}+x, \frac{1}{4}+y, \frac{1}{4}-z; \frac{1}{4}+z, \frac{1}{4}+x, \frac{1}{4}-y; \frac{1}{4}+y, \frac{1}{4}+z, \frac{1}{4}-x;$ $\frac{1}{4}-x, \frac{1}{4}-z, \frac{1}{4}-y; \frac{1}{4}-y, \frac{1}{4}-x, \frac{1}{4}-z; \frac{1}{4}-z, \frac{1}{4}-y, \frac{1}{4}-x;$ $\frac{1}{4}-x, \frac{1}{4}+z, \frac{1}{4}+y; \frac{1}{4}-y, \frac{1}{4}+x, \frac{1}{4}+z; \frac{1}{4}-z, \frac{1}{4}+y, \frac{1}{4}+x;$ $\frac{1}{4}+x, \frac{1}{4}-z, \frac{1}{4}+y; \frac{1}{4}+y, \frac{1}{4}-x, \frac{1}{4}+z; \frac{1}{4}+z, \frac{1}{4}-y, \frac{1}{4}+x;$ $\frac{1}{4}+x, \frac{1}{4}+z, \frac{1}{4}-y; \frac{1}{4}+y, \frac{1}{4}+x, \frac{1}{4}-z; \frac{1}{4}+z, \frac{1}{4}+y, \frac{1}{4}-x.$	General: $hkl: h+k, k+l, (l+h)=2n$ $hhl: (l+h=2n); \bar{C}$ $0kl: (k, l=2n); k+l=4n$ \bar{C}
96	<i>h</i> 2	$\frac{1}{8}, x, \frac{1}{4}-x; \frac{1}{8}, \frac{1}{4}-x, x; \frac{3}{8}, x, \frac{1}{2}+x; \frac{3}{8}, \frac{1}{4}+x, x;$ $\frac{1}{4}-x, \frac{1}{8}, x; x, \frac{1}{8}, \frac{1}{4}-x; \frac{1}{4}+x, \frac{3}{8}, x; x, \frac{3}{8}, \frac{1}{4}+x;$ $x, \frac{1}{4}-x, \frac{1}{8}; \frac{1}{4}-x, x, \frac{1}{8}; x, \frac{1}{4}+x, \frac{3}{8}; \frac{1}{4}+x, x, \frac{3}{8};$ $\frac{1}{8}, \bar{x}, \frac{3}{4}+x; \frac{1}{8}, \frac{3}{4}+x, \bar{x}; \frac{3}{8}, \bar{x}, \frac{3}{4}-x; \frac{3}{8}, \frac{3}{4}-x, \bar{x};$ $\frac{3}{4}+x, \frac{1}{8}, \bar{x}; \bar{x}, \frac{1}{8}, \frac{3}{4}+x; \frac{3}{4}-x, \frac{3}{8}, \bar{x}; \bar{x}, \frac{3}{8}, \frac{3}{4}-x;$ $\bar{x}, \frac{3}{4}+x, \frac{1}{8}; \frac{3}{4}+x, \bar{x}, \frac{1}{8}; \bar{x}, \frac{3}{4}-x, \frac{3}{8}; \frac{3}{4}-x, \bar{x}, \frac{3}{8}.$	Special: as above, plus
96	<i>g</i> <i>m</i>	$x, x, z; z, x, x; x, z, x; \bar{x}, x, \bar{z}; \bar{z}, x, \bar{x}; \bar{x}, z, \bar{x};$ $x, \bar{x}, \bar{z}; z, \bar{x}, \bar{x}; x, \bar{z}, \bar{x}; \bar{x}, \bar{x}, z; \bar{z}, \bar{x}, x; \bar{x}, \bar{z}, x;$ $\frac{1}{4}-x, \frac{1}{4}-x, \frac{1}{4}-z; \frac{1}{4}-z, \frac{1}{4}-x, \frac{1}{4}-x; \frac{1}{4}-x, \frac{1}{4}-z, \frac{1}{4}-x;$ $\frac{1}{4}-x, \frac{1}{4}+x, \frac{1}{4}+z; \frac{1}{4}-z, \frac{1}{4}+x, \frac{1}{4}+x; \frac{1}{4}-x, \frac{1}{4}+z, \frac{1}{4}+x;$ $\frac{1}{4}+x, \frac{1}{4}-x, \frac{1}{4}+z; \frac{1}{4}+z, \frac{1}{4}-x, \frac{1}{4}+x; \frac{1}{4}+x, \frac{1}{4}-z, \frac{1}{4}+x;$ $\frac{1}{4}+x, \frac{1}{4}+x, \frac{1}{4}-z; \frac{1}{4}+z, \frac{1}{4}+x, \frac{1}{4}-x; \frac{1}{4}+x, \frac{1}{4}+z, \frac{1}{4}-x.$	no extra conditions
48	<i>f</i> <i>mm</i>	$x, 0, 0; \bar{x}, 0, 0; \frac{1}{4}+x, \frac{1}{4}, \frac{1}{4}; \frac{1}{4}-x, \frac{1}{4}, \frac{1}{4};$ $0, x, 0; 0, \bar{x}, 0; \frac{1}{4}, \frac{1}{4}+x, \frac{1}{4}; \frac{1}{4}, \frac{1}{4}-x, \frac{1}{4};$ $0, 0, x; 0, 0, \bar{x}; \frac{1}{4}, \frac{1}{4}, \frac{1}{4}+x; \frac{1}{4}, \frac{1}{4}, \frac{1}{4}-x.$	$hkl: h+k+l=2n+1$ or $4n$
32	<i>e</i> <i>3m</i>	$x, x, x; \frac{1}{4}-x, \frac{1}{4}-x, \frac{1}{4}-x;$ $x, \bar{x}, \bar{x}; \frac{1}{4}-x, \frac{1}{4}+x, \frac{1}{4}+x;$ $\bar{x}, x, \bar{x}; \frac{1}{4}+x, \frac{1}{4}-x, \frac{1}{4}+x;$ $\bar{x}, \bar{x}, x; \frac{1}{4}+x, \frac{1}{4}+x, \frac{1}{4}-x.$	no extra conditions
16	<i>d</i> <i>3m</i>	$\frac{5}{8}, \frac{5}{8}, \frac{5}{8}; \frac{5}{8}, \frac{7}{8}, \frac{7}{8}; \frac{7}{8}, \frac{5}{8}, \frac{7}{8}; \frac{7}{8}, \frac{7}{8}, \frac{5}{8}.$	$hkl: \begin{cases} h=2n+1 \\ k=2n+1 \\ l=2n+1 \end{cases}$ $\begin{cases} 4n+2 \\ \text{or } 4n+2 \end{cases} \begin{cases} 4n \\ \text{or } 4n \end{cases}$ $4n+2 \quad 4n$
16	<i>c</i> <i>3m</i>	$\frac{1}{8}, \frac{1}{8}, \frac{1}{8}; \frac{1}{8}, \frac{3}{8}, \frac{3}{8}; \frac{3}{8}, \frac{1}{8}, \frac{3}{8}; \frac{3}{8}, \frac{3}{8}, \frac{1}{8}.$	
8	<i>b</i> <i>43m</i>	$\frac{1}{2}, \frac{1}{2}, \frac{1}{2}; \frac{3}{4}, \frac{3}{4}, \frac{3}{4}.$	
8	<i>a</i> <i>43m</i>	$0, 0, 0; \frac{1}{4}, \frac{1}{4}, \frac{1}{4}.$	$hkl: h+k+l=2n+1$ or $4n$

Exercise 3

NaAlSi₂O₆·H₂O (analcite, zeolite) cubic (Fig. 3s, see solution)

- **Chemical composition:** NaAlSi₂O₆·H₂O - **Cell parameter:** $a = 13.73 \text{ \AA}$ - **Density:** $\delta = 2.28 \text{ gr/cm}^3$ - **Atomic weights:** Na 22.991, Al 26.982, Si 28.086, O 15.999.
- **Space group:** hkl diffractions occur only if $h+k+l = 2n$; therefore, the lattice is cI ; besides, OkI occur only for k and l even and hhl occur only for $2h+l = 4n$; from morphology the point group of analcrite is $m\bar{3}m$, thus the space group is $Ia\bar{3}d$. Assign atoms to the Wyckoff positions.

	Number of positions, Wyckoff notation, and point symmetry	Co-ordinates of equivalent positions	Conditions limiting possible reflections
		$(0,0,0; \frac{1}{2}, \frac{1}{2}, \frac{1}{2}) +$	
			General:
96	h 1	$x, y, z; \frac{1}{2} + x, \frac{1}{2} - y, \bar{z}; \bar{x}, \frac{1}{2} + y, \frac{1}{2} - z; \frac{1}{2} - x, \bar{y}, \frac{1}{2} + z;$ $z, x, y; \frac{1}{2} + z, \frac{1}{2} - x, \bar{y}; z, \frac{1}{2} + x, \frac{1}{2} - y; \frac{1}{2} - z, \bar{x}, \frac{1}{2} + y;$ $y, z, x; \frac{1}{2} + y, \frac{1}{2} - z, \bar{x}; \bar{y}, \frac{1}{2} + z, \frac{1}{2} - x; \frac{1}{2} - y, \bar{z}, \frac{1}{2} + x;$ $\bar{x}, \bar{y}, \bar{z}; \frac{1}{2} - x, \frac{1}{2} + y, z; x, \frac{1}{2} - y, \frac{1}{2} + z; \frac{1}{2} + x, y, \frac{1}{2} - z;$ $\bar{z}, \bar{x}, \bar{y}; \frac{1}{2} - z, \frac{1}{2} + x, y; z, \frac{1}{2} - x, \frac{1}{2} + y; \frac{1}{2} + z, x, \frac{1}{2} - y;$ $\bar{y}, \bar{z}, \bar{x}; \frac{1}{2} - y, \frac{1}{2} + z, x; y, \frac{1}{2} - z, \frac{1}{2} + x; \frac{1}{2} + y, z, \frac{1}{2} - x;$ $\frac{1}{4} + x, \frac{1}{4} + z, \frac{1}{4} + y; \frac{1}{4} + y, \frac{1}{4} + x, \frac{1}{4} + z; \frac{1}{4} + z, \frac{1}{4} + y, \frac{1}{4} + x;$ $\frac{3}{4} + x, \frac{1}{4} - z, \frac{3}{4} - y; \frac{3}{4} + y, \frac{1}{4} - x, \frac{3}{4} - z; \frac{3}{4} + z, \frac{1}{4} - y, \frac{3}{4} - x;$ $\frac{3}{4} - x, \frac{3}{4} + z, \frac{1}{4} - y; \frac{3}{4} - y, \frac{3}{4} + x, \frac{1}{4} - z; \frac{3}{4} - z, \frac{3}{4} + y, \frac{1}{4} - x;$ $\frac{1}{4} - x, \frac{3}{4} - z, \frac{3}{4} + y; \frac{1}{4} - y, \frac{3}{4} - x, \frac{3}{4} + z; \frac{1}{4} - z, \frac{3}{4} - y, \frac{3}{4} + x;$ $\frac{1}{4} - x, \frac{1}{4} - z, \frac{1}{4} - y; \frac{1}{4} - y, \frac{1}{4} - x, \frac{1}{4} - z; \frac{1}{4} - z, \frac{1}{4} - y, \frac{1}{4} - x;$ $\frac{3}{4} - x, \frac{1}{4} + z, \frac{3}{4} + y; \frac{3}{4} - y, \frac{1}{4} + x, \frac{3}{4} + z; \frac{3}{4} - z, \frac{1}{4} + y, \frac{3}{4} + x;$ $\frac{3}{4} + x, \frac{3}{4} - z, \frac{1}{4} + y; \frac{3}{4} + y, \frac{3}{4} - x, \frac{1}{4} + z; \frac{3}{4} + z, \frac{3}{4} - y, \frac{1}{4} + x;$ $\frac{1}{4} + x, \frac{3}{4} + z, \frac{1}{4} - y; \frac{1}{4} + y, \frac{3}{4} + x, \frac{1}{4} - z; \frac{1}{4} + z, \frac{3}{4} + y, \frac{1}{4} - x.$	$hkl: h+k+l=2n$ $hhl: (l=2n); 2h+l=4n; \odot$ $OkI: k, (l)=2n; \odot$
			Special: as above, plus
48	g 2	$\frac{1}{8}, x, \frac{1}{4} - x; \frac{3}{8}, x, \frac{3}{4} + x; \frac{5}{8}, \bar{x}, \frac{1}{4} - x; \frac{7}{8}, \bar{x}, \frac{3}{4} + x;$ $\frac{1}{4} - x, \frac{1}{8}, x; \frac{3}{4} + x, \frac{3}{8}, x; \frac{1}{4} - x, \frac{5}{8}, \bar{x}; \frac{3}{4} + x, \frac{7}{8}, \bar{x};$ $x, \frac{1}{4} - x, \frac{1}{8}; x, \frac{3}{4} + x, \frac{3}{8}; \bar{x}, \frac{1}{4} - x, \frac{5}{8}; \bar{x}, \frac{3}{4} + x, \frac{7}{8};$ $\frac{1}{8}, \bar{x}, \frac{1}{4} + x; \frac{3}{8}, \bar{x}, \frac{3}{4} - x; \frac{5}{8}, x, \frac{1}{4} + x; \frac{7}{8}, x, \frac{3}{4} - x;$ $\frac{1}{4} + x, \frac{1}{8}, \bar{x}; \frac{3}{4} - x, \frac{3}{8}, \bar{x}; \frac{1}{4} + x, \frac{5}{8}, x; \frac{3}{4} - x, \frac{7}{8}, x;$ $\bar{x}, \frac{1}{4} + x, \frac{1}{8}; \bar{x}, \frac{3}{4} - x, \frac{3}{8}; x, \frac{1}{4} + x, \frac{5}{8}; x, \frac{3}{4} - x, \frac{7}{8}.$	$hkl: h \text{ or } k \text{ or } l=2n+1$ or $4n$
48	f 2	$x, 0, \frac{1}{4}; \bar{x}, \frac{1}{2}, \frac{1}{4}; \frac{1}{4} - x, 0, \frac{1}{4}; \frac{3}{4} + x, \frac{1}{2}, \frac{1}{4};$ $\frac{1}{4}, x, 0; \frac{1}{4}, \bar{x}, \frac{1}{2}; \frac{1}{4}, \frac{1}{4} - x, 0; \frac{1}{4}, \frac{3}{4} + x, \frac{1}{2};$ $0, \frac{1}{4}, x; \frac{1}{2}, \frac{1}{4}, \bar{x}; 0, \frac{1}{4}, \frac{1}{4} - x; \frac{1}{2}, \frac{1}{4}, \frac{3}{4} + x;$ $x, \frac{1}{2}, \frac{3}{4}; \bar{x}, 0, \frac{3}{4}; \frac{1}{4} - x, \frac{1}{2}, \frac{3}{4}; \frac{3}{4} + x, 0, \frac{3}{4};$ $\frac{3}{4}, x, \frac{1}{2}; \frac{3}{4}, \bar{x}, 0; \frac{3}{4}, \frac{1}{4} - x, \frac{1}{2}; \frac{3}{4}, \frac{3}{4} + x, 0;$ $\frac{1}{2}, \frac{3}{4}, x; 0, \frac{3}{4}, \bar{x}; \frac{1}{2}, \frac{3}{4}, \frac{1}{4} - x; 0, \frac{3}{4}, \frac{3}{4} + x.$	$hkl: \text{ If } h, k, (l)=2n,$ then $h+k+l=4n$
32	e 3	$x, x, x; \frac{1}{2} + x, \frac{1}{2} - x, \bar{x}; \bar{x}, \frac{1}{2} + x, \frac{1}{2} - x; \frac{1}{2} - x, \bar{x}, \frac{1}{2} + x;$ $\bar{x}, \bar{x}, \bar{x}; \frac{1}{2} - x, \frac{1}{2} + x, x; x, \frac{1}{2} - x, \frac{1}{2} + x; \frac{1}{2} + x, x, \frac{1}{2} - x;$ $\frac{1}{4} + x, \frac{1}{4} + x, \frac{1}{4} + x; \frac{1}{4} - x, \frac{1}{4} - x, \frac{1}{4} - x;$ $\frac{3}{4} + x, \frac{1}{4} - x, \frac{3}{4} - x; \frac{3}{4} - x, \frac{1}{4} + x, \frac{3}{4} + x;$ $\frac{3}{4} - x, \frac{3}{4} + x, \frac{1}{4} - x; \frac{3}{4} + x, \frac{3}{4} - x, \frac{1}{4} + x;$ $\frac{1}{4} - x, \frac{3}{4} - x, \frac{3}{4} + x; \frac{1}{4} + x, \frac{3}{4} + x, \frac{3}{4} - x.$	$hkl: \text{ If } h, k, (l)=2n,$ then $h+k+l=4n$
24	d 4	$\frac{3}{8}, 0, \frac{1}{4}; \frac{1}{4}, \frac{3}{8}, 0; 0, \frac{1}{4}, \frac{3}{8}; \frac{1}{8}, 0, \frac{3}{4}; \frac{3}{4}, \frac{3}{8}, 0; 0, \frac{3}{4}, \frac{1}{4};$ $\frac{7}{8}, 0, \frac{1}{4}; \frac{1}{4}, \frac{7}{8}, 0; 0, \frac{1}{4}, \frac{7}{8}; \frac{5}{8}, 0, \frac{3}{4}; \frac{3}{4}, \frac{7}{8}, 0; 0, \frac{3}{4}, \frac{5}{8}.$	$hkl: \left. \begin{array}{l} h=2n+1 \\ k=2n+1 \\ l=4n+2 \end{array} \right\} \begin{array}{l} 4n+2 \\ \text{or } 4n+2 \\ 4n \end{array}$ $\left. \begin{array}{l} 8n \\ \text{or } 8n+4 \\ 4n+2 \end{array} \right\} \begin{array}{l} 4n \\ \text{or } 4n \\ 4n \end{array}$ $h, k, l \text{ permutable}$
24	c 222	$\frac{1}{8}, 0, \frac{1}{4}; \frac{1}{4}, \frac{1}{8}, 0; 0, \frac{1}{4}, \frac{1}{8}; \frac{3}{8}, 0, \frac{3}{4}; \frac{3}{4}, \frac{3}{8}, 0; 0, \frac{3}{4}, \frac{3}{8};$ $\frac{7}{8}, 0, \frac{1}{4}; \frac{1}{4}, \frac{7}{8}, 0; 0, \frac{1}{4}, \frac{7}{8}; \frac{5}{8}, 0, \frac{3}{4}; \frac{3}{4}, \frac{5}{8}, 0; 0, \frac{3}{4}, \frac{5}{8}.$	$hkl: \left. \begin{array}{l} h=2n+1 \\ k=2n+1 \\ l=4n+2 \end{array} \right\} \begin{array}{l} 4n \\ \text{or } 4n \\ 4n \end{array}$ $h, k, l \text{ permutable}$
16	b 32	$\frac{1}{8}, \frac{1}{8}, \frac{1}{8}; \frac{3}{8}, \frac{3}{8}, \frac{3}{8}; \frac{7}{8}, \frac{7}{8}, \frac{7}{8}; \frac{5}{8}, \frac{5}{8}, \frac{5}{8};$ $\frac{7}{8}, \frac{7}{8}, \frac{1}{8}; \frac{3}{8}, \frac{3}{8}, \frac{1}{8}; \frac{1}{8}, \frac{3}{8}, \frac{5}{8}; \frac{5}{8}, \frac{1}{8}, \frac{3}{8}.$	$hkl: \left. \begin{array}{l} h=2n+1 \\ k=2n+1 \\ l=4n+2 \end{array} \right\} \begin{array}{l} 4n \\ \text{or } 4n \\ 4n \end{array}$ $h, k, l \text{ permutable}$
16	a $\bar{3}$	$0, 0, 0; 0, \frac{1}{2}, \frac{1}{2}; \frac{1}{2}, 0, \frac{1}{2}; \frac{1}{2}, \frac{1}{2}, 0;$ $\frac{1}{4}, \frac{1}{4}, \frac{1}{4}; \frac{1}{4}, \frac{3}{4}, \frac{3}{4}; \frac{3}{4}, \frac{1}{4}, \frac{3}{4}; \frac{3}{4}, \frac{3}{4}, \frac{1}{4}.$	$hkl: h, k, (l)=2n;$ $h+k+l=4n$

Exercise 4

Kalifersite and the palysepiole polysomatic series (Fig. 1e and 4s, see solution)

Kalifersite - $(K,Na)_5(Fe^{3+})_7[Si_{20}O_{50}](OH)_6 \cdot 12H_2O$; $P\bar{1}$, $a = 14.86$, $b = 20.54$, $c = 5.29$ Å, $\alpha = 95.6$, $\beta = 92.3$, $\gamma = 94.4^\circ$. Structure unknown.

Sepiolite - $Mg_8[Si_{12}O_{30}](OH)_4 \cdot 12H_2O$; $Pncn$, $a = 13.40$, $b = 26.80$, $c = 5.28$ Å. Structure known (Fig. 1e).

Palygorskite - $Mg_5[Si_8O_{20}](OH)_2 \cdot 8H_2O$; $C2/m$, $a = 13.27$, $b = 17.868$, $c = 5.279$ Å, $\beta = 107.38^\circ$. Structure known (Fig. 1e).

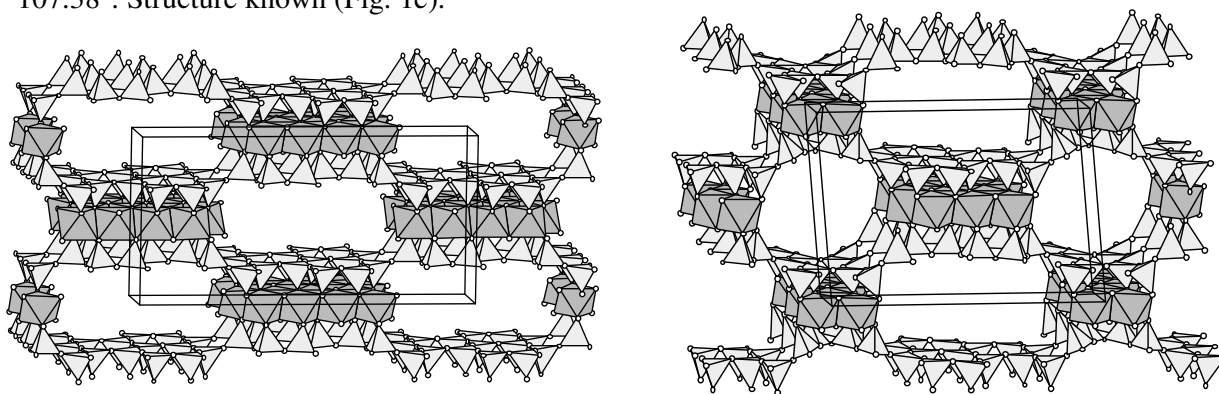


Figure 1e - Sepiolite (left) and palygorskite (right).

The following is noted

1. Kalifersite, sepiolite and palygorskite have close values of their a and c parameters. The $[001]$ direction corresponds to the fibre axis of these silicates and its periodicity corresponds to that of a pyroxene chain.
2. The b value of kalifersite is intermediate between that of palygorskite and sepiolite.
3. The $[Si_{20}O_{50}](OH)_6$ silicate anion of kalifersite corresponds to the sum of those of sepiolite, $[Si_{12}O_{30}](OH)_4$, and palygorskite, $[Si_8O_{20}](OH)_2$.
4. The structures of sepiolite and palygorskite are based on a framework of interconnected $[001]$ TOT ribbons which correspond to cuts with different width of the phyllosilicate 2:1 layer (Fig. 3 and 4). These ribbons are chess-board arranged and delimit channels. In the $[010]$ direction, the $(TOT)_S$ ribbon of sepiolite is one chain wider than that, $(TOT)_P$, of palygorskite. That requires for sepiolite a b value about 9 Å longer than in palygorskite, *i.e.* about 4.5 Å per T chain. Mg and H_2O are in the channels delimited by the silicate ribbons.

Taking into account the above chemical and crystallographic aspects derive a structure model for kalifersite.

Exercise 5

Modelling the structure of seidite-(Ce) (Fig. 2e and 5s, see solution)

Seidite-(Ce) - $\text{Na}_4(\text{Ce},\text{Sr})_2\text{Ti}[(\text{Si}_8\text{O}_{18})(\text{OH})_2](\text{O},\text{OH},\text{F})_4 \cdot 5\text{H}_2\text{O}$; $a = 24.61$, $b = 7.23$, $c = 14.53$ Å, $\beta = 94.6^\circ$; $C2/c$. Structure is unknown.

Rhodesite - $\text{K}_2\text{Ca}_4[\text{Si}_8\text{O}_{18}(\text{OH})_2] \cdot 12\text{H}_2\text{O}$; $a = 23.416$, $b = 6.555$, $c = 7.050$ Å, $Pm\bar{m}$. The structure is known (Fig. 2e). Several compounds share with rhodesite the silicate double layer crossed by eight-membered channels delimited by Si-tetrahedra. The structures of these compounds differ instead in the content of the octahedral layer that is sandwiched between two silicate modules. In rhodesite, Ca is in the octahedral layer arranged in chains of edge-sharing Ca octahedra; K and H_2O are in the channels.

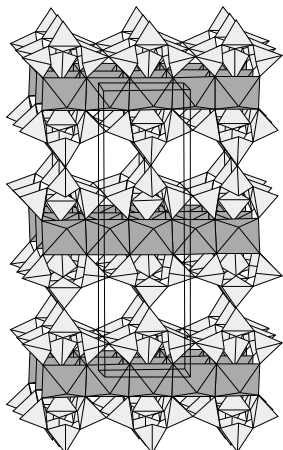


Figure 2e – The crystal structure of rhodesite.

The following is noted.

1. Seidite-(Ce) and rhodesite share the anion $[(\text{Si}_8\text{O}_{18})(\text{OH})_2]$ which corresponds to the composition of the double layer mentioned above.
2. The two minerals show close values of the cell parameters [$c/2$ seidite-(Ce) to be compared with c of rhodesite].
- 3.

Table 1 – Members of the rhodesite group.

Name	Chemical formula	a, b, c (Å), β (°)
Seidite-(Ce)	$\text{Na}_4(\text{Ce},\text{Sr})_2\{\text{Ti}(\text{OH})_2(\text{Si}_8\text{O}_{18})\}(\text{O},\text{OH},\text{F})_4 \cdot 5\text{H}_2\text{O}$	24.61, 7.23, 14.53, 94.6
Rhodesite	$\text{K}_2\text{Ca}_4[\text{Si}_8\text{O}_{18}(\text{OH})_2] \cdot 12\text{H}_2\text{O}$	23.416, 6.555, 7.050
Macdonaldite	$\text{BaCa}_4[\text{Si}_8\text{O}_{18}(\text{OH})_2] \cdot 10\text{H}_2\text{O}$	14.081, 13.109, 23.560
Delhayelite	$\text{K}_7\text{Na}_3\text{Ca}_5[\text{Si}_7\text{AlO}_{19}]_2\text{F}_4\text{Cl}_2$	24.86, 7.07, 6.53
Hydrodelhayelite	$\text{K}_2\text{Ca}_4[\text{Si}_7\text{AlO}_{17}(\text{OH})_2]_2 \cdot 6\text{H}_2\text{O}$	6.648, 23.846, 7.073
Monteregianite-(Y)	$\text{K}_2\text{Na}_4\text{Y}_2[\text{Si}_8\text{O}_{19}]_2 \cdot 10\text{H}_2\text{O}$	9.512, 23.956, 9.617, 93.85